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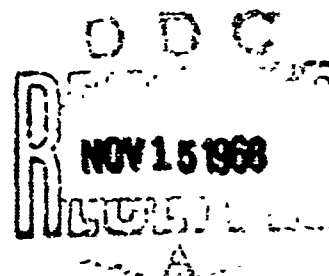
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## CARBIDES FOR SOLID PROPELLANT NOZZLE SYSTEMS

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TECHNICAL REPORT AFRPL-TR-68-164

1 OCTOBER 1968



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## CARBIDES FOR SOLID PROPELLANT NOZZLE SYSTEMS

F. T. Lally  
D. P. Lavery

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## FOREWORD

This report is the final technical report summarizing the results of the work performed under Contract FO 4611-67-C-0094 covering the period of 1 October 1967 to 30 June 1968. The progress made during the final month of the program (1 June 1968 to 30 June 1968) is also included in this report. This report has been assigned the TRW Internal Report Number ER 7307.

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Air Force Rocket Propulsion Laboratory

Research and Technology Division

Edwards, California

Air Force Systems Command

United States Air Force

Lt. David Zorich acted as Project Engineer.

This technical report has been reviewed and is approved.

Lt. David Zorich  
Project Engineer  
Air Force Rocket Propulsion Laboratory

### ABSTRACT

The magnitude of reactivity between  $\text{Al}_2\text{O}_3$  and carbide composite rocket nozzle throat materials was assessed by means of laboratory tests. A secondary objective of the program was to establish the validity of the laboratory tests in evaluating material performance. The program included a plasma jet test that measured mechanical and chemical erosion and a static reactivity test to separate the purely chemical effects. A prediction of the performance of the carbide composites in a test firing was made, based on the laboratory tests.

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## INTRODUCTION

The performance of rocket nozzle inserts with relatively small throat diameters is often dependent upon maintaining the dimensional integrity of the nozzle throat section during the firing cycle. The primary degradation mechanisms involve melting, thermal shock, and the combined effect of chemical reactivity and mechanical erosion. Recent work has indicated that carbides are susceptible to surface degradation by chemical reactivity when exposed to alumina ( $Al_2O_3$ ) at temperatures substantially below the carbide melting points. As a result, the limiting consideration in selecting the high melting carbides for Aluminum containing solid propellant rocket nozzle application may be chemical reactivity rather than melting point.

Due to the complexity of the degradation mechanisms operating in the nozzle throat during the firing cycle, it is desirable to isolate the effects of the individual mechanisms by a laboratory test. The subject program was aimed at establishing the magnitude of the reactivity problem between  $Al_2O_3$  and carbide composites being considered for rocket nozzle throat materials. A secondary objective of the program was to establish the validity of the laboratory tests as a tool for evaluating nozzle throat material performance potential.

The program included a plasma-jet test that subjected the specimen to high temperatures and impingement of  $Al_2O_3$  particles to measure the combined effects of mechanical erosion and chemical erosion. The effect of chemical reactivity alone was evaluated by a static reactivity (sessile drop) test wherein the selected composite material was heated to 4500°F or higher in contact with molten  $Al_2O_3$ .

## MATERIALS AND TEST PROCEDURE

### Test Material

The material evaluated during this program consisted of four distinct carbide composites furnished by the Government. The first and second sets of materials were Ta-Hf-C microcomposites produced by the Aerojet-General Corporation by hot pressing. The materials are described in report Nos. AFRPL-TR-66-282, AFRPL-TR-67-13, and AFRPL-TR-67-207. The chemical composition of these materials were identified by Aerojet-General as 8Ta-55Hf-37C and 8Ta-54Hf-38C.

The third and fourth sets of carbide materials tested in this program were also received from Aerojet-General Corporation. These specimens, which were not identified as to specific composition, are hypereutectic TaC-graphite composites produced by fusion casting and containing varying percentages of graphite. The fourth carbide composite which was to be TaC clad and hence tested in the unmachined condition. The samples were tested with the square cross section as furnished by Aerojet. All other specimens were machined to a 0.250 inch diameter prior to testing.

### Test Procedure

Plasma-jet impingement tests - The apparatus for conducting the plasma-jet impingement tests is shown in the schematic diagram of Figure 1. The specimen, a 1/4 inch diameter x 2 inch long rod is resistance heated by 220V powerstats which are capable of developing temperatures up to 6000°F. The plasma-jet is attached to a movable arm which allows the gun to be rotated into the plane of the specimen. The erosion resistance is measured as a function of temperature by maintaining a fixed gun distance and independently varying the temperature of the resistance heating (preheat). An argon environment was used in the chamber with an alumina injection rate of 0.5 lbs./hr. and a heat flux from the gun of approximately 1200 BTU/ft<sup>2</sup>-sec. Temperatures were measured during the test with an optical pyrometer. The test duration was three minutes and the change in diameter was used as the evaluation parameter.

Static reactivity tests - The static reactivity was measured by sessile drop tests which were carried out with a 1/4 inch diameter rod type specimen heated by resistance to the desired temperature. A cavity, machined in the top of the specimen, was filled with Al<sub>2</sub>O<sub>3</sub> powder. The specimen had the least cross section at the bottom of the cavity and therefore reached its highest temperature in this area. These tests were carried out for a duration of 3 minutes in an atmosphere of flowing argon.

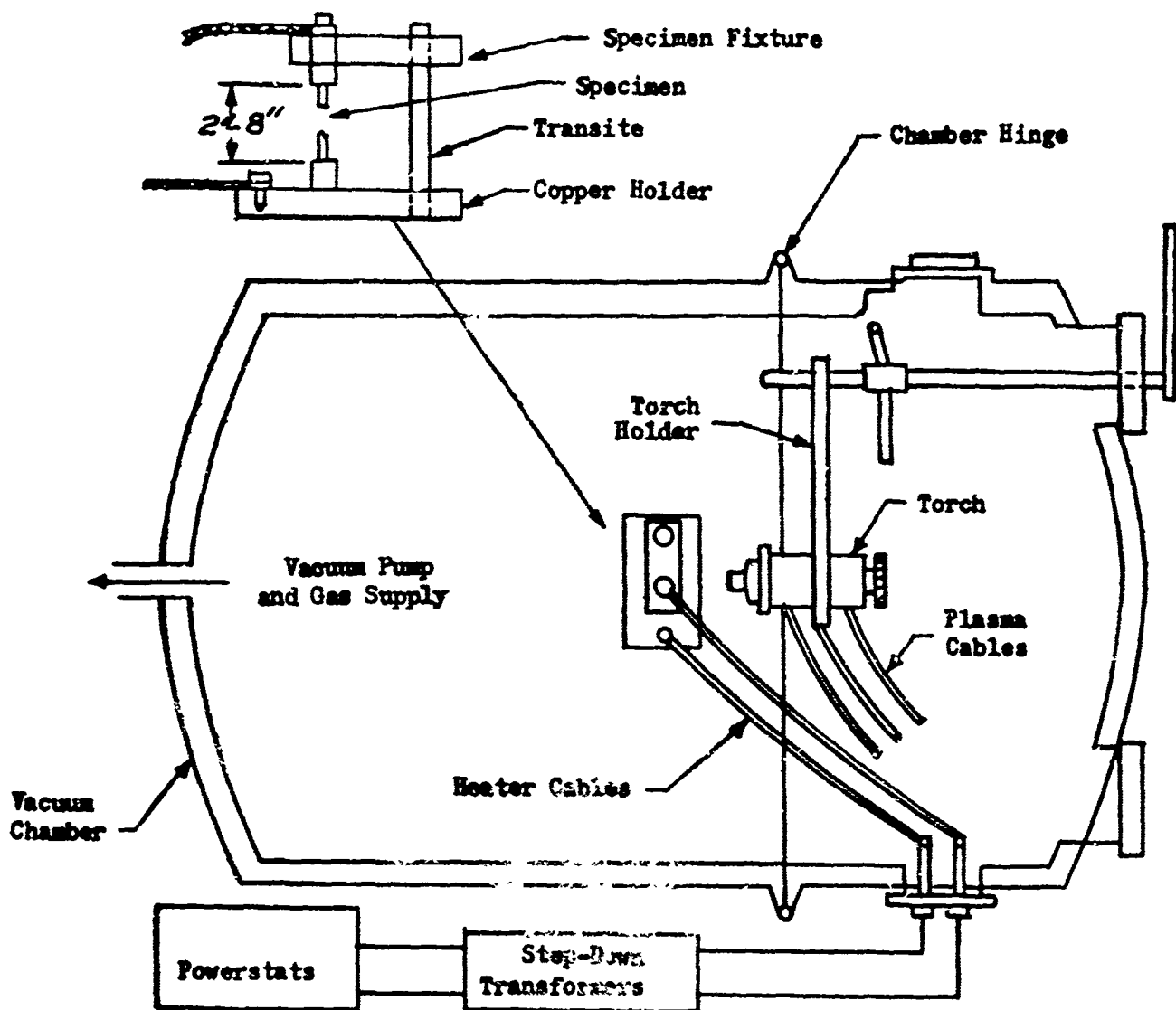


Figure 1. Schematic Diagram of Plasma Jet Impingement Test Apparatus

Specimen evaluation - The main evaluation criterion in these tests was the depth of surface reactivity or erosion experienced by the specimen during the test. In addition to erosion measurements, the specimens were evaluated by metallographic examination, electroprobe, and x-ray diffraction to assess the nature and extent of chemical reactions and to define the mechanisms of attack.

## TEST RESULTS

The results of both the plasma jet impingement tests and the static reactivity tests on the two sets of microcomposites materials will be presented first, since these two sets of specimens behaved similarly to each other. The results obtained on the hyperautactitic carbides will then be presented so that an overall comparison of the various materials can be made.

### Microcomposites

Plasma jet impingement tests - The first specimen of the 8Ta-55Hf-37C composition was brought up to a preheat temperature of 4250°F by resistance heating in approximately 15 seconds. The specimen failed catastrophically (apparently by melting) before the plasma gun could be rotated into place. Specimen No. 2 was then preheated at a much slower rate, requiring 18 minutes to reach 5600°F, and then cooled to room temperature. This specimen failed during the cool-down cycle due to mechanical stresses imposed by the clamping mechanism. All remaining specimens that required preheating were brought up to temperature at a rate of 1000°F per minute. The appearance of the 8Ta-54Hf-37C microcomposite specimens after testing are presented in Figure 2; the 8Ta-55Hf-38C material is shown in Figure 3.

The data derived from the plasma-jet impingement tests of both microcomposites materials are tabulated in Table 1. The erosion rate is also plotted as a function of test temperature in Figure 4. The erosion rates of other refractory nozzle materials are included in the graph of Figure 4 for comparison.

In general, the erosion resistance of both of the microcomposites materials in the alumina seeded plasma jet tests is comparable to that of stoichiometric HfC at temperatures of 5400°F and higher. Below this temperature, the data indicated a somewhat lower erosion rate for the microcomposite materials. The results also indicated that, in common with the stoichiometric HfC, a threshold temperature between 5400-5600°F exists above which the recession rate increases rapidly.

Static reactivity of microcomposites The conditions of the static reactivity tests of the microcomposites materials are tabulated in Table 2. (The disposition of the 20 microcomposite specimens received from the Government is also included in Table 2.)

The specimens, after testing, are shown in Figure 5. The fracture of specimen No. 3 occurred after 2.2 minutes at 4800°F and was accompanied by severe evidence of deformation of the specimen. All specimens in this test showed some evidence of deformation, although not as severe as that of specimen No. 3.

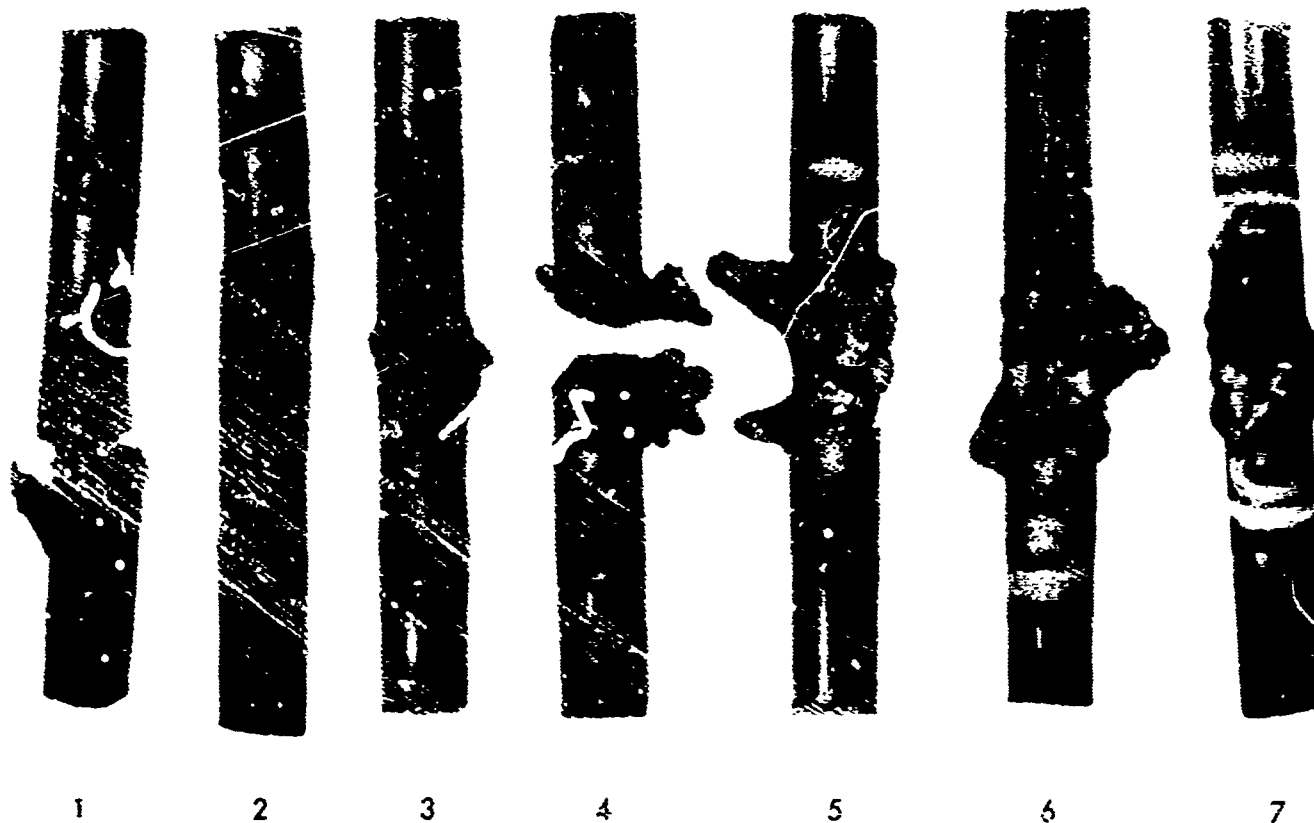


Figure 2. Appearance of 8Ta-55Hf-37C Microcomposites After Exposure to  $\text{Al}_2\text{O}_3$  Seeded Plasma Jet Tests. The Specimen Number Identity Can Be Correlated with Test Conditions Given in Table 1.



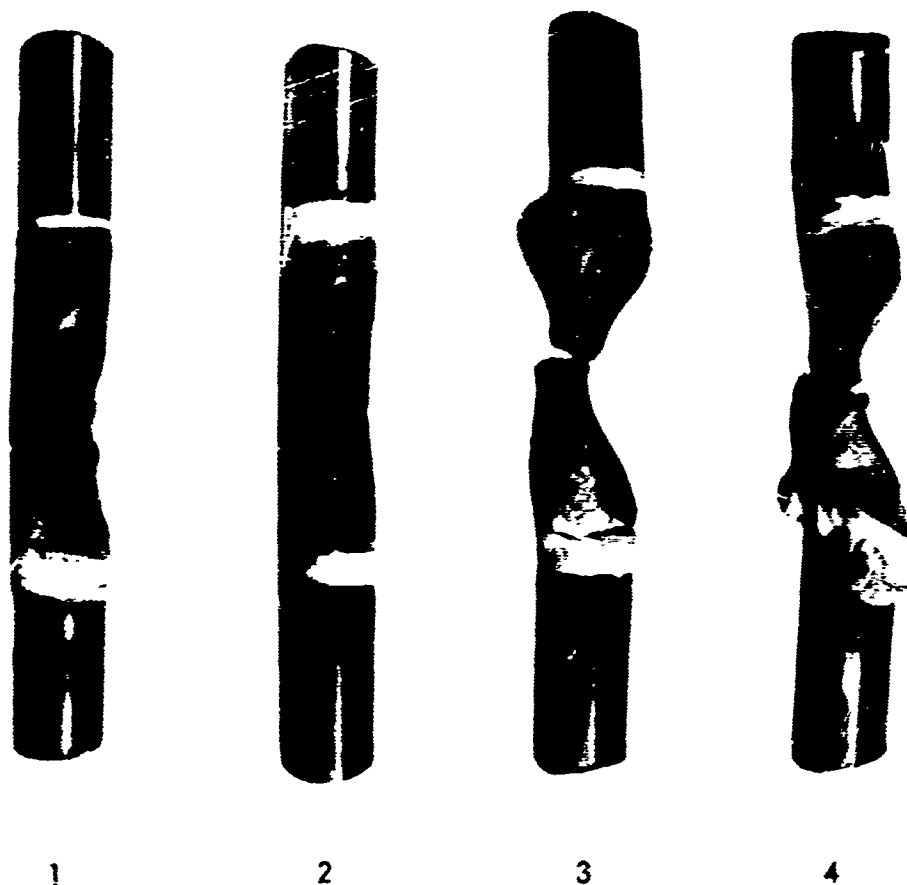


Figure 3. Appearance of 81a-54Hf-38C Microcomposites After Exposure to  $\text{Al}_2\text{O}_3$  Seeded Plasma Jet Tests. The Specimen Number Can Be Correlated with Test Conditions Listed in Table 1

TABLE 1

Plasma Jet Impingement Data, Al<sub>2</sub>O<sub>3</sub> Seeded

<u>Specimen No.</u>	<u>Preheat Temperature °F</u>	<u>Test Temperature °F</u>	<u>Recession Rate inches/min.</u>
8Ta-55Hf-37C Composition			
3	4050	5400	0.035
4	3720	5450	0.096
5	3000	5550	0.074
6	2300	5400	0.038
7	None	5050	0.018

8Ta-54Hf-38C Composition

1	1500	5800	0.010
2	None	5500	0.030
3	1500	5400	0.060
4	1500	5600	0.040

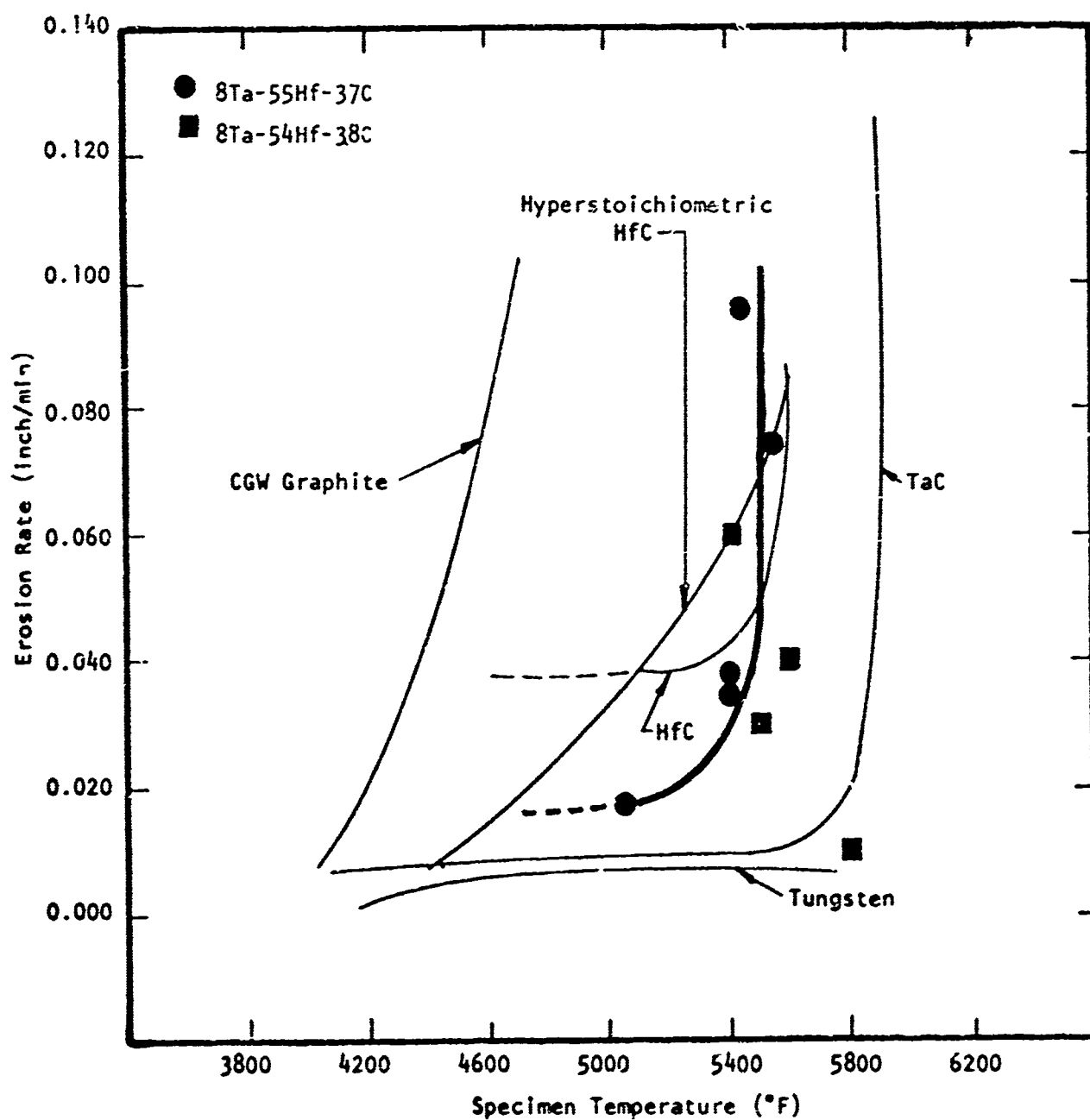


Figure 4. Erosion Rates of Ta-Hf-C Microcomposites Compared to Various Nozzle Materials, Plasma Jet Impingement Tests, Alumina Seeded, 0.5 lbs/hr.

TABLE 2

Test Conditions for the Alumina-Microcomposite Carbide

Static Reactivity Tests

<u>Specimen No.</u>	<u>Composition</u>	<u>Test Temperature °F</u>	<u>Test Duration Minutes</u>
1	8Ta-54Hf-38C	4600	3.0
2	8Ta-54Hf-38C	5100	3.0
3	8Ta-54Hf-38C	4800	2.2
4	8Ta-55Hf-37C	5300	3.0

Disposition of Microcomposite Specimens Received

	<u>8Ta-55Hf-37C</u>	<u>8Ta-54Hf-38C</u>
Plasma Jet Test	7	4
Metallography (as-received)	1	1
Static Reactivity	1	3
Broken Before Testing	1	2
Totals	<u>10</u>	<u>10</u>



Figure 5. Appearance of 8Ta-54Hf-38C Microcomposites After Exposure to  $Al_2O_3$  in the Static Reactivity Tests

In all tests, the alumina readily wet the microcomposite surface and flowed out of the cavity covering the entire heated surface of the specimen. After testing, the specimens were evaluated for the extent and mechanism of chemical reaction with the alumina by use of metallography, electron microprobe analysis, and x-ray diffraction. The results of these examinations are discussed later in this report.

#### Hypereutectic TaC-C Composites

Plasma-jet impingement tests - The data derived from the plasma jet impingement tests of the hypereutectic TaC-C carbide composites are tabulated in Table 3. Specimens of the third group, after testing, are presented in Figure 6. A cycle time of 3 minutes at temperature was maintained except for specimens Nos. 3 and 6, Figure 6. Specimen No. 3 eroded completely through after 2 minutes. The test of specimen No. 6 was aborted after 30 seconds by failure of the plasma jet nozzle which resulted in thermal stress failure of the specimen. The fourth group of specimens shown in Figure 7 were all tested with a 3 minute cycle time.

The erosion rate of both carbide materials is plotted as a function of test temperature in Figure 8. The erosion rates of other refractory nozzle materials as well as the microcomposites tested previously are included in the graph of Figure 8 for comparison.

The erosion resistance of both of the hypereutectic carbide specimens to the alumina seeded plasma flame are comparable and difficult to separate. The data are also characterized by a large degree of scatter. The erosion resistance as a function of temperature is lower than that of the Ta-Hf-C microcomposite materials tested in this program. A threshold temperature such as was found to exist for the microcomposite materials was not sharply defined for the hypereutectic carbides.

Static reactivity tests - The static reactivity tests were performed under the same conditions as were used in testing the microcomposite materials. The conditions of testing the TaC-C composites are tabulated in Table 4.

In all tests, the alumina readily wet the carbide surface and flowed out of the cavity covering the entire heated surface of the specimen. Visual examination of the specimens after the test showed no evidence of gross damage, and heated surfaces were bright and clean.

TABLE 3

Plasma Jet Impingement Data, Al<sub>2</sub>O<sub>3</sub> Seeded

3rd Group of Carbide Composites

<u>Identification No.</u>	<u>Aerojet Specimen No.</u>	<u>Preheat Temp. °F</u>	<u>Test Temp. °F</u>	<u>Recession Rate in/min.</u>
1	463-b	None	4650	.023
2	463-d	1500	4760	.038
3	463-e	1925	5030	.125
4	463-j	None	5090	.014
5	463-c	1600	5250	.083
6	463-h	3100	5610	.116

4th Group of Carbide Composites

1	487-6	None	4550	.004
2	487-5	3000	4870	.083
3	487-4	None	4920	.026
4	487-2	None	5200	.018
5	487-3	1500	5300	.057
6	487-1	None	5640	.021

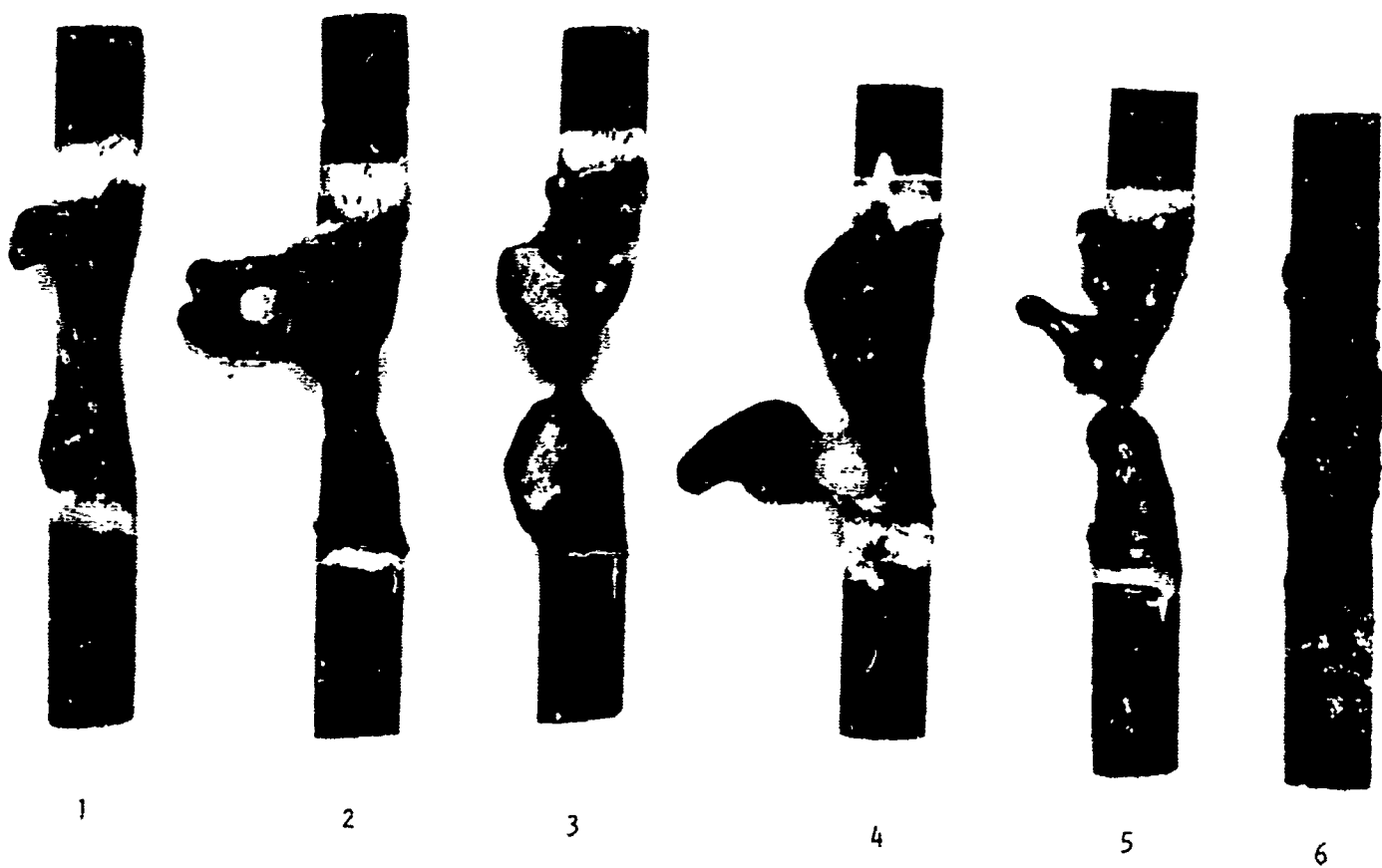


Figure 6. Appearance of the 3rd Group of Carbide Composites After Exposure to  $\text{Al}_2\text{O}_3$  Seeded Plasma Jet Tests.



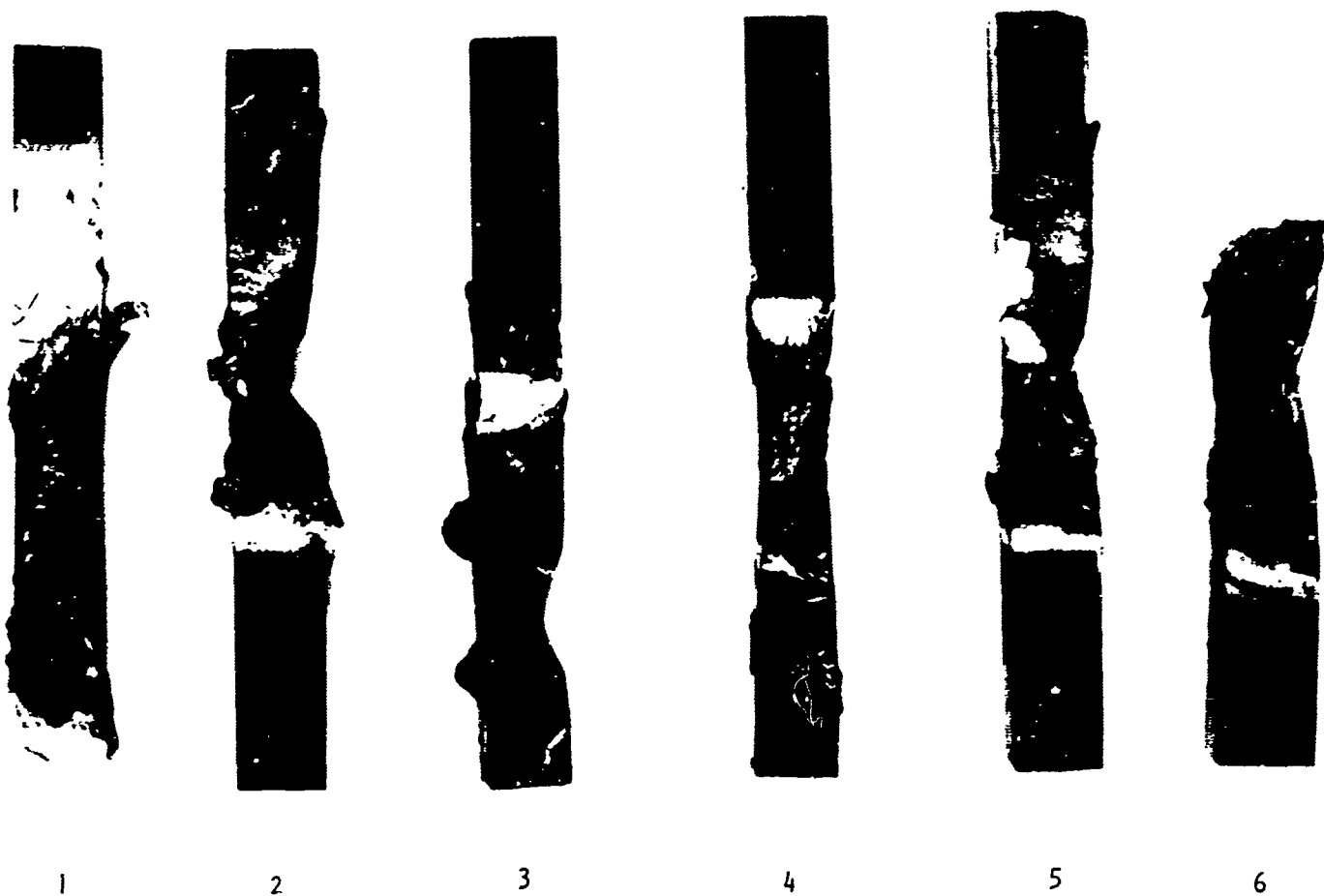


Figure 7. Appearance of the 4th Group of Carbide Composites After Exposure to  $\text{Al}_2\text{O}_3$  Seeded Plasma Jet Tests

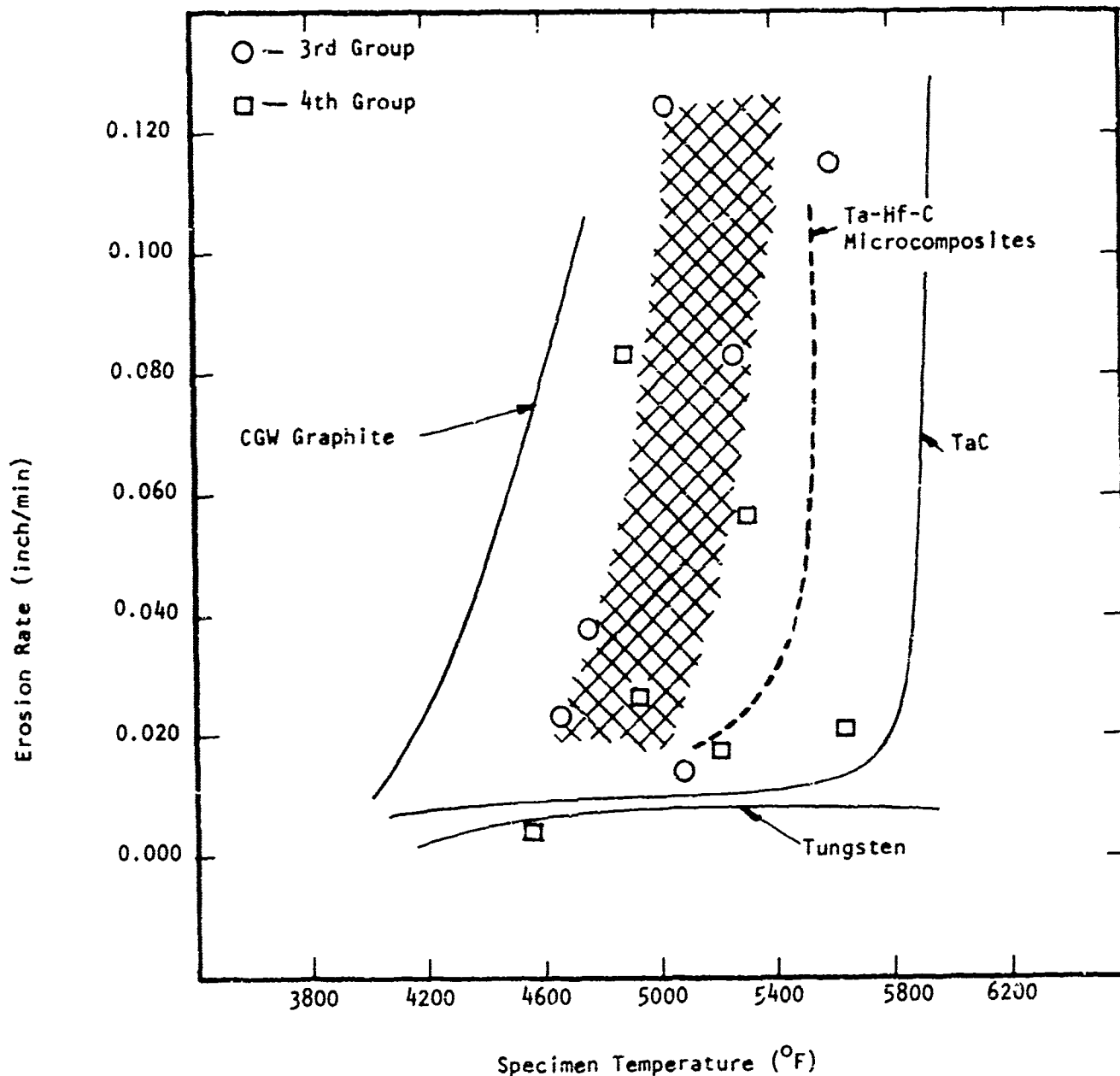


Figure 8. Erosion Rates of Hypereutectic TaC-C Carbides Compared to Various Nozzle Materials During Impingement with a Plasma Jet Seeded with Alumina at 0.5 lbs./hr.

TABLE 4

Static Reactivity (Sessile Drop) Test-Al<sub>2</sub>O<sub>3</sub>--

3rd Group of Carbide Composites

<u>Identification No.</u>	<u>Aerojet Specimen No.</u>	<u>Test Temperature °F</u>	<u>Test Duration Minutes</u>
1	463-f	4650	3
2	462-h	4800	3
3	462-c	4850	3
4	462-j	5300	1.5

4th Group of Carbide Composites

1	488-1	Broken before testing
2	488-2	4750 3
3	488-4	Broken before testing
4	488-3	Broken before testing

### Metallurgical Analysis of Specimens

After the carbide specimens were subjected to either the plasma jet impingement test or the static reactivity test, selected specimens were examined by various metallurgical techniques to evaluate the nature and mechanism of chemical reactivity with the alumina. This analysis included metallography, electron microprobe analysis, and x-ray diffraction analysis.

Metallography - Typical microstructures of the microcomposite materials both as-received and after plasma jet testing are shown in Figures 9 and 10. The photomicrograph of Figure 10 indicates considerable penetration of alumina into the specimen. The typical reaction was an intergranular attack. This is illustrated in Figure 11, a sessile drop specimen of the 8Ta-54Hf-38C composition heated to 4600°F. In this specimen, attack by the alumina penetrated to a depth of 60 mils below the surface of the specimen. No evidence of general melting was found on this specimen. This degree of attack and resulting microstructure is typical of all the microcomposite specimens tested (including the 8Ta-55Hf-37C specimens). At the higher temperatures, it might be expected that attack would be more severe. However, the wetting of the entire specimen surface by the alumina at over 5000°F proceeded so rapidly that the alumina rapidly left the specimen cavity and was therefore unable to support localized attack.

The static reactivity specimen tested at 4800°F underwent more severe reaction and showed indications that significant molten material had been present at the test temperature. It is presumed that the molten phase was a eutectic formed by the carbide composite and the alumina. Metallographic examination of this specimen after testing, however, showed a microstructure similar to that of the other specimens, as shown in Figure 11. One explanation of the accelerated reaction at 4800°F is that, up to this temperature, the rate of chemical attack increased faster than the ability of the alumina to escape from the cavity by its increased wettability. Therefore, the alumina remained concentrated in one area of the specimen to support the localized reaction.

The as-received hypereutectic TaC-C composites were characterized by a variety of microstructures as shown in Figure 12. The predominant microstructure, however, is the carbide matrix with finely distributed graphite and containing relatively massive graphite plates illustrated in Figure 12b. The microstructure of typical specimens subjected to the Al<sub>2</sub>O<sub>3</sub> seeded plasma jet test are presented in Figure 13. In contrast to the microcomposites, the attack on the hypereutectic carbides is not intergranular but proceeds along the paths occupied by the massive graphite plates. As will be shown later, the single phase area shown in Figure 13a is tantalum oxide formed in the location presumable formerly occupied by a graphite flake.

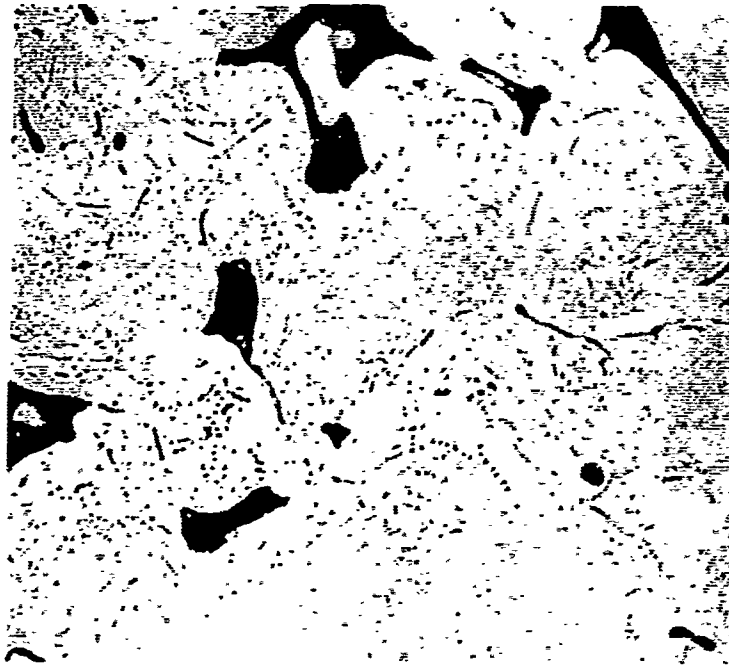


Figure 9. Microstructure of the as-received microcomposite. 500X

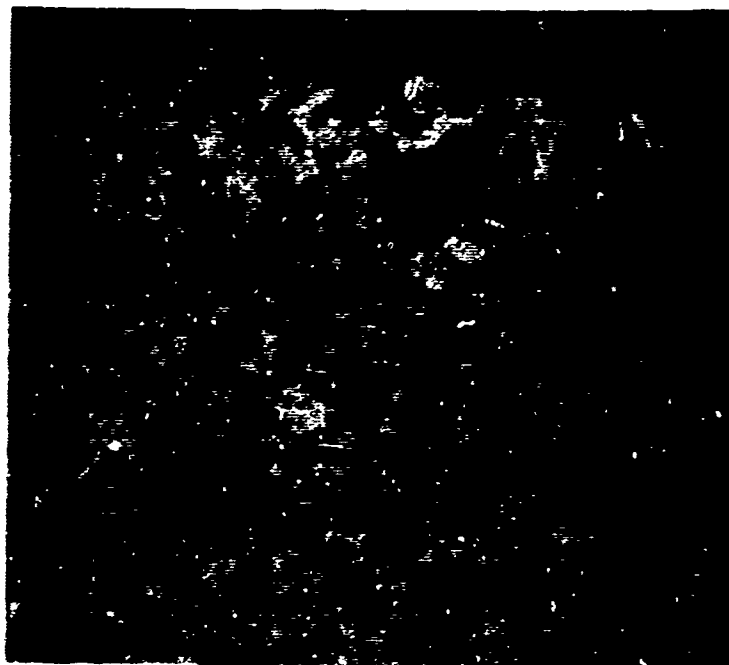
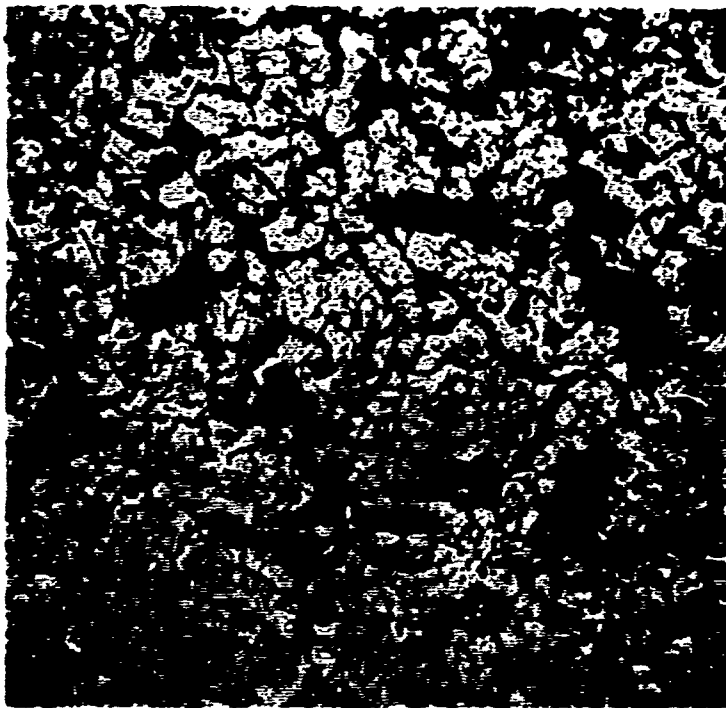
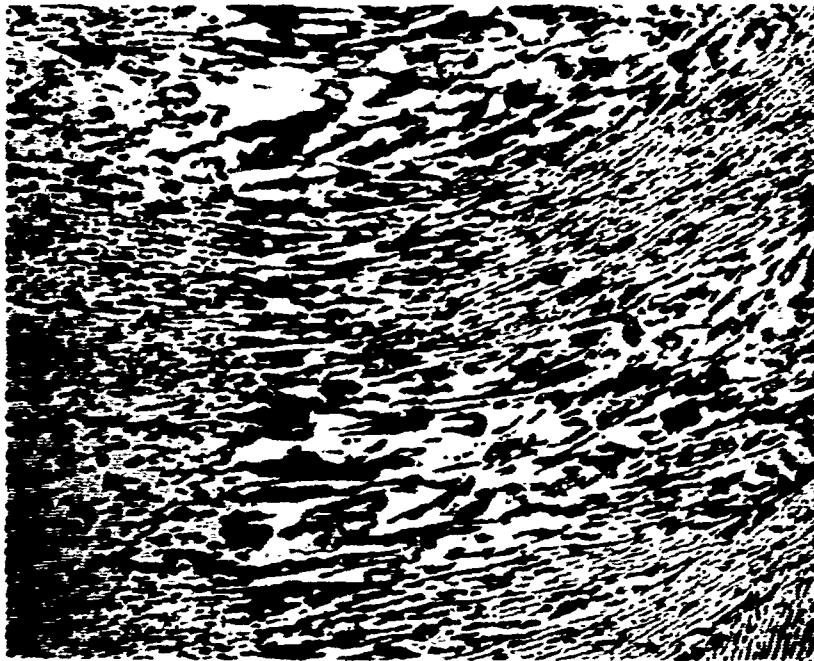


Figure 10. Microcomposite material after the alumina seeded plasma jet test at 5600°F. 250X



500X

Figure 11. Microstructure of the 8Ta-54Hf-38C Microcomposite  
After a 3 Minute Exposure at 4600°F in Contact  
With  $\text{Al}_2\text{O}_3$ .



(a)

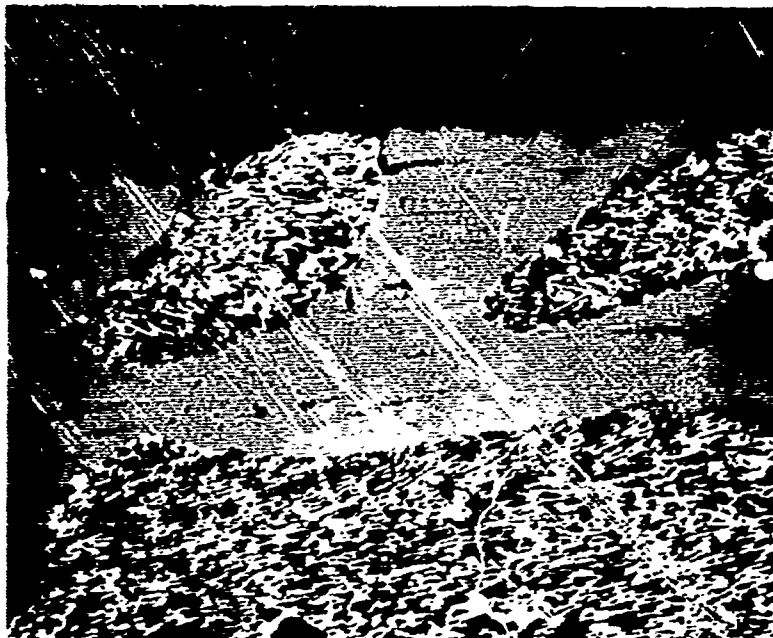
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(b)

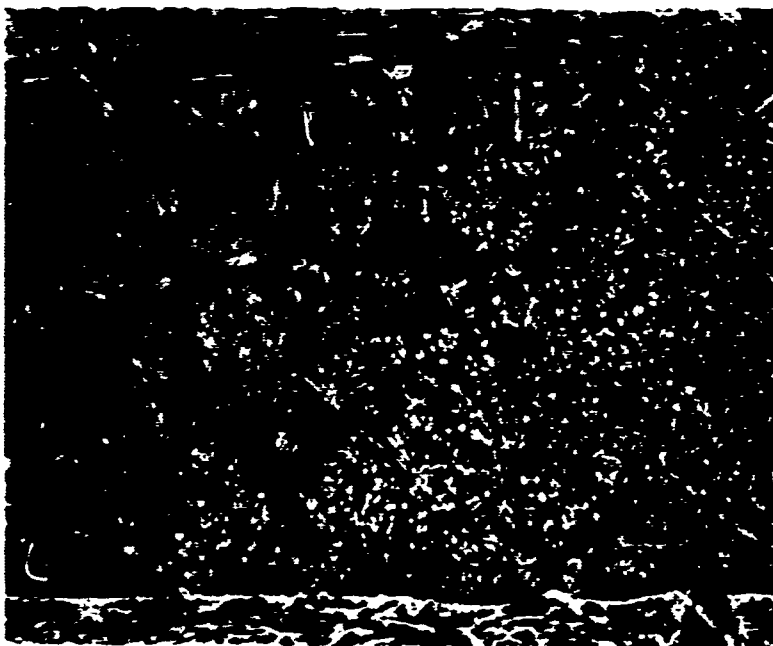
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Figure 12. Microstructure of As-Received TaC-C Hypereutectic Composite. Aerojet Specimen No. 462J



(a) Aerojet Specimen No. 463b  
Plasma Jet Tested for 3 Minutes at 4650°F

250X



(b) Aerojet Specimen No. 487-1  
Plasma Jet Tested for 3 Minutes at 5640°F

250X

Figure 13. Hypereutectic TaC-C Composites after the  $Al_2O_3$  Seeded Plasma Jet Test. For Aid in Interpreting Microstructure, See Microprobe Analysis in Figure 17.



Electron microprobe analysis - Electron microprobe analyses were performed on these specimens in order to help define the mechanism of attack on the carbide composite by the alumina. Electron microprobe photos of the reacted area of the 8Ta-54Hf-38C microcomposite, specimen No. 4 - Table 1, after plasma jet testing at 5600°F, are shown in Figure 14. The photographs show that the attack is proceeding primarily at the expense of the hafnium (by the correspondence of the Hf and the Al x-ray patterns). This same mechanism of attack is indicated in the static reactivity test specimens. A typical set of results (for specimen No. 1, 8Ta-54Hf-38C microcomposite, 4600°F exposure) are shown in Figure 15. These photographs indicate the following:

1. Segregation of Ta is taking place
2. Localized areas of aluminum are evident with some relation to the oxygen distribution (indicating  $Al_2O_3$ )
3. A possible indication of attack of the HfC by  $Al_2O_3$  based on the similar distributions of the Hf, C, and oxygen patterns

The results of the specimen tested at 5100°F show these same trends, and in a less subtle manner. These results, presented in the series of photographs in Figure 16, indicate the following:

1. Segregation of the Hf and Ta: the high intensity (light) areas in the Ta and the Hf images are not coincident.
2. Presence of oxides of Al and Hf: areas of high oxygen and aluminum roughly corresponding to areas rich in hafnium
3. Relatively uniform carbon distribution but somewhat higher in areas of high oxygen indicating chemical attack by  $Al_2O_3$ , is on HfC rather than on metallic constituents.

Typical sets of results of the electron microprobe analysis of the hypereutectic carbides are shown in Figures 17 and 18. The analysis of specimen 463b - Table 3 shown in Figure 17 was performed at the point of greatest erosion within the body of the specimen. The analysis shown in Figure 18, was performed on the same specimen but at the midpoint within the reaction product which had built up on the O.D. during the plasma jet test. The photograph of Figure 17 shows a wide distribution of Ta and  $O_2$  with a faint but discernable pattern, almost no Al, and a distribution pattern of C. The coincident patterns of Ta and  $O_2$  and the low level of C in the single phase region, coupled with the virtual absence of Al, show the single phase region to be composed of tantalum oxide.



Hf X-Rays 400X



Ta X-Rays 400X

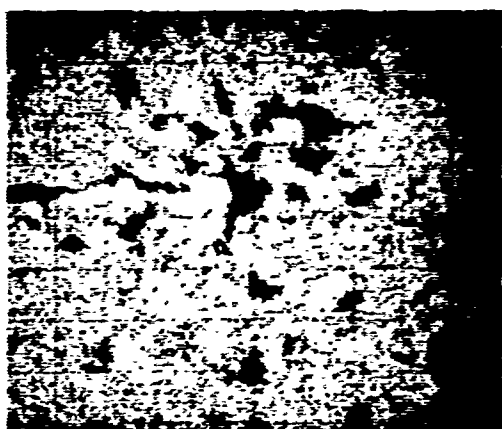


Al X-Rays 400X



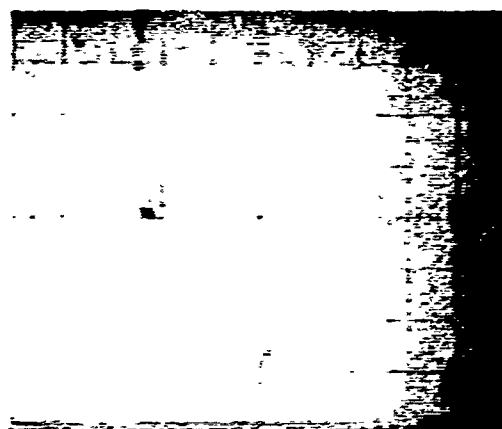
Sample Current 400X

Figure 14. Electron Microprobe Photographs of the Microcomposite Material After the Alumina Seeded Plasma Jet Test at 5600°F



Ta X-rays

400X



Hf X-rays

400X



C X-rays

400X



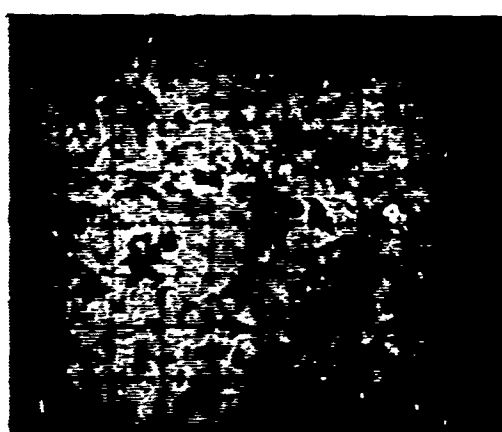
O X-rays

400X



Al X-rays

400X



Sample Current

400X

Figure 15. Electron Microprobe Photographs of the 8Ta-54Hf-38C Microcomposite after Static Reactivity Testing at 4600°F in Contact with  $Al_2O_3$



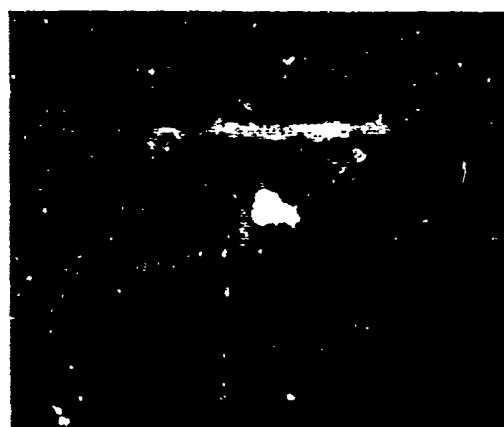
Ta X-rays

400X



Hf X-rays

400X



C X-rays

400X



O X-rays

400X



Al X-rays

400X



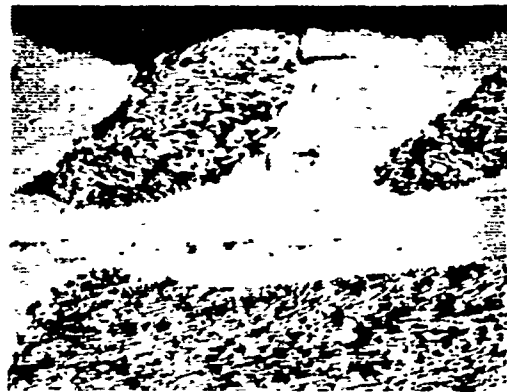
Sample Current

400X

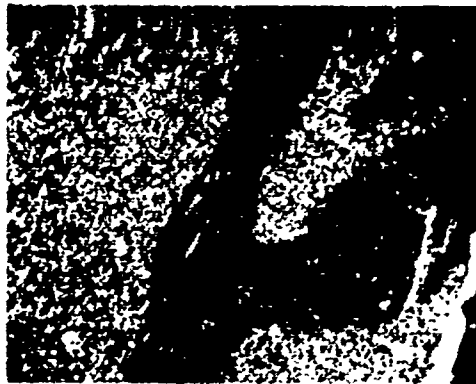
Figure 16. Electron Microprobe Photographs of the 8Ta-54Hf-35C Microcomposites after Static Reactivity Testing at 5100°F in Contact with  $Al_2O_3$



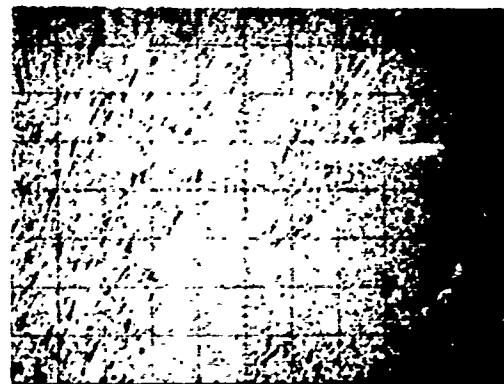
Secondary Current 300X



Photomicrograph 250X



C X-Rays 300X



Ta X-Rays 300X

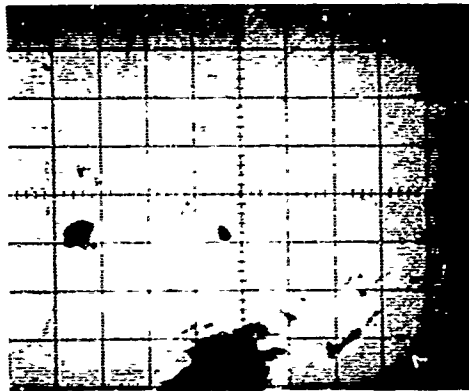


Al X-Rays 300X



O<sub>2</sub> X-Rays 300X

Figure 17. Electron Microprobe Photographs of the Reacted Zone within the TaC-C Composite, Specimen 463b - Table 3, after The Alumina Seeded Plasma Jet Test at 4650°F/



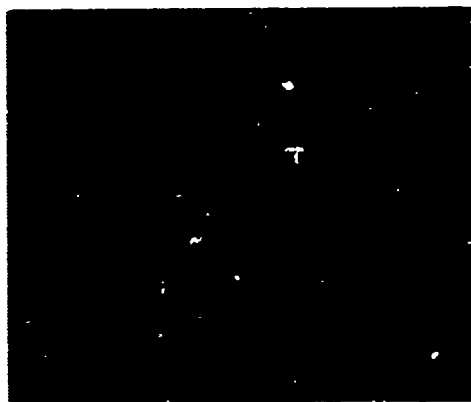
Secondary Current 300X



Al X-Rays 300X



Ta X-Rays 300X



C X-Rays 300X



O<sub>2</sub> X-Rays 300X

Figure 18. Electron Microprobe Photographs of the Area within The Build Up Reaction Product of the TaC-C Composite Specimen 463b - Table 3, After the Alumina Seeded Plasma Jet Test at 4650°F

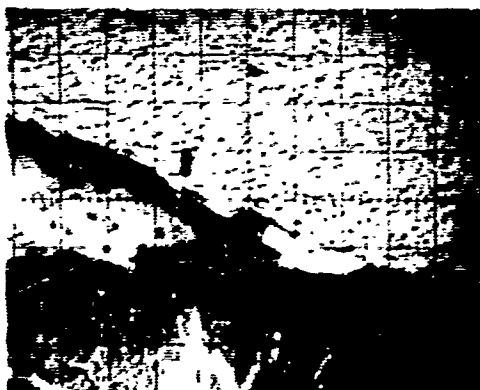
In the analysis of the reaction product (Figure 18), the uniform distribution of Ta and  $O_2$  and the low level of C and Al indications, show the reaction product to be tantalum oxide. A qualitative analysis of the reaction product (2  $\mu$  scan) made with the electron microprobe in the same area shown in Figure 18, revealed the major element to be Ta with a trace of Al.

The initial stages of the reaction are presented in Figure 19, an analysis performed on the static reactivity specimen No. 488-2 - Table 3 after a 3 minute exposure at 4750°F in the sessile drop test. The photographs show a high concentration of C in the alumina build up and also the penetration of alumina along the channel once occupied by a graphite plate.

X-ray diffraction - X-ray diffraction studies were made on the as-received material and on the static reactivity tested material. Measurements were made using a Norelco Diffractometer with  $CuK\alpha$  radiation and Ni filter.

The as-received microcomposite materials were primarily FCC  $\lambda$  (HfTa)C with lattice parameter of 4.606 Å. The x-ray diffraction pattern also showed a strong line for metallic Hf (d spacing (101) of 2.42 Å). The static reactivity tested material showed that two distinct carbides are present; both are FCC  $\lambda$  (HfTa)C with lattice parameters of 4.591 Å and 4.580 Å. This material also yielded a strong line for HCP Hf with a (101) d spacing of 2.40 Å. This indicates some dissolved Ta in the metallic Hf. The one apparent explanation that is consistent with the phase diagram data is that Hf is selectively removed from the matrix carbide on exposure, thus causing a localized lattice parameter reduction.

X-ray diffraction studies were also made on typical specimens of the plasma jet and sessile drop hypereutectic carbide materials. The x-ray diffraction patterns indicated the major constituent to be TaC with varying (slight to appreciable) quantities of the high temperature form of  $Ta_2O_5$ . Indications of an unknown compound (which could be a non-stoichiometric tantalum oxide) were also found. The presence of  $Al_2O_3$  was not detected by the x-ray diffraction analysis on any of the samples tested.



Secondary Current 300X



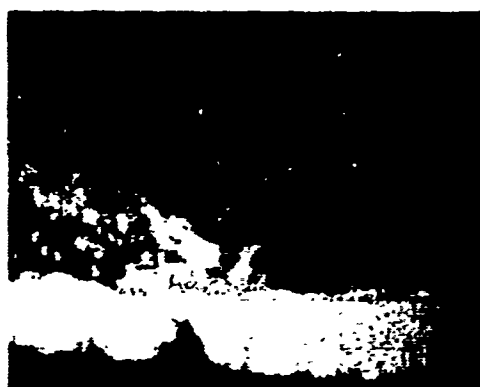
Microstructure 250X



C X-Rays 300X



Ta X-Rays 300X



Al X-Rays 300X



O<sub>2</sub> X-Rays 300X

Figure 19. Electron Microprobe Photographs of the TaC-C Composite, (Specimen No 488-2 - Table 4), after Static Reactivity Testing at 4750°F



## DISCUSSION OF RESULTS

The plasma jet impingement tests and the static reactivity tests performed in this program show that significant differences exist in the response to these tests by the two major types - microcomposites and hypereutectic TaC-C - of carbide materials evaluated. As shown in Figure 8, both types of material show erosion behavior in the general range defined by previous tests of other graphitic and carbide-type materials. Figure 8 also shows that the hypereutectic material can be expected to experience a given erosion rate at several hundred degrees (°F) lower temperature than the microcomposites.

There was no significant difference in behavior noted for the various material compositions within the two main classes of carbide materials. The compositional differences between the two microcomposites was slight. It should be noted that the compositions of the hypereutectic carbides were not supplied to TRW. The difference between the two sets of hypereutectic carbide samples supposedly was a TaC coating on the one set. No coating was evident, however, so it is thought these two sets of specimens may have been essentially identical.

In addition to the actual difference in the erosion rates between the microcomposite and the hypereutectic carbides, the more significant difference is in the mechanism of attack. Understanding the mechanism of reaction is the first step in being able to extrapolate the data to actual rocket firing conditions and in developing realistic approaches for improved materials. The aspects of the reaction of  $Al_2O_3$  with these types of carbides as brought to light in this program are summarized below:

### Microcomposites

The mechanism responsible for the accelerated erosion rate of the microcomposite carbides appears to be a eutectic reaction between the liquid alumina particles and hafnium in the microcomposite. The electron microprobe analysis shows that the attack is proceeding primarily at the expense of the hafnium. The reaction between HfC and liquid  $Al_2O_3$  has been found to be exothermic in previous tests in this laboratory which would explain the rapid increase in erosion above 5400°F.

The exothermic reaction would also explain the reason that pre-heat temperatures have so little influence on the final test temperature achieved by the plasma jet. As the data in Table I indicate, there is little correlation between preheat temperature and test temperature even though power input to the plasma torch was constant throughout the series of tests. The electron microprobe examination also indicates segregation of tantalum and hafnium has occurred which allows the composite material to act in a similar manner to HfC during the course of the plasma jet test.

The behavior of the microcomposite material in the static reactivity test is similar to its behavior in the plasma jet impingement test. The degradation mechanism is chemical in nature, and involves an exothermic eutectic reaction of  $\text{Al}_2\text{O}_3$  and  $\text{HfC}$ . The reaction mechanism proceeds more rapidly in the plasma impingement test because the impinging  $\text{Al}_2\text{O}_3$  particles provide a continuous intimate contact of the reactive components. The sweeping action of the jet also serves to remove the reacted material to permit the reaction to continue.

The temperature at which the reaction rate becomes very significant can be as low as approximately  $4800^\circ\text{F}$ , as was indicated by specimen No. 3, Figure 5. The relatively undamaged appearance of specimen No. 2 ( $5100^\circ\text{F}$  exposure), Figure 5 is probably due to the fact that most of the  $\text{Al}_2\text{O}_3$ , extremely fluid at the test temperature of  $5100^\circ\text{F}$ , flowed out of the pocket leaving too little material at the hottest portion of the specimen to permit the reaction to become damaging.

The plasma jet impingement test demonstrated that the maximum use temperature for the microcomposite materials in contact with  $\text{Al}_2\text{O}_3$  is below  $5400^\circ\text{F}$ . The static reactivity tests indicated that the useful temperature limit of these materials may be as low as  $4800^\circ\text{F}$ . The usefulness of these microcomposites as rocket nozzle throat insert materials in contact with  $\text{Al}_2\text{O}_3$  at surface temperatures between  $4800$  and  $5400^\circ\text{F}$  will likely be dependent on specific conditions of use, such as gas pressure, alumina content of exhaust, and required length of the duty cycle.

#### Hypereutectic Carbides

The erosion of the TaC-C hypereutectic composites in the  $\text{Al}_2\text{O}_3$  seeded plasma-jet tests results from a chemical reaction aided by mechanical erosion from the plasma jet stream. The chemical reaction probably proceeds by a reduction of alumina by carbon and the subsequent formation of tantalum oxide. The absence of Al in most of the electron microprobe photographs is due to vaporization of the Al as a metal or in the form of a carbide.

The speed of the reaction depends on the presence of liquid alumina and free carbon in sufficient quantities to sustain the reaction. The random, but not necessarily uniform, distribution of graphite plates in the hypereutectic microstructure would then account for the scatter of data points observed in the  $\text{Al}_2\text{O}_3$  seeded plasma jet tests.

## CONCLUSIONS AND PREDICTION OF FIRING BEHAVIOR

A secondary objective of the subject program is to establish the validity of the  $\text{Al}_2\text{O}_3$  seeded plasma jet test in conjunction with the static reactivity test as a laboratory tool for evaluation of rocket nozzle throat materials. A prediction of the performance of the composite carbides in two test firings to be conducted under Contract AF04(611)-11608 are made using data derived from the laboratory tests together with an outline of the test firing conditions and an estimate of the nozzle wall temperature furnished by the Government.

The composite materials are to be test fired on the AFRPL 40-inch diameter uncured propellant test motor under the following conditions: burn time of 60 seconds; estimated motor MEOP of 700 psi; propellant aluminum content of 27%.

The thermal map provided by the government indicates a predicted temperature of  $4960^\circ\text{F}$  at the hottest point of the insert wall at  $T=5$  seconds. Extrapolation of these data indicates that the throat insert wall will reach a temperature of  $5400^\circ\text{F}$  shortly before  $T=25$  seconds.

The plasma jet and static reactivity test data indicates that  $5400^\circ\text{F}$  is the threshold temperature for the microcomposite materials. Above this temperature, the erosion rate becomes catastrophic because of a eutectic reaction between the liquid alumina and the hafnium in the microcomposite.

The threshold temperature for the TaC-C hypereutectic composites was not sharply defined by the plasma jet tests. The threshold temperature is, however, below  $5400^\circ\text{F}$ . Consequently, the laboratory test data indicated a greater degree of erosion for the hypereutectic composites.

The plasma jet and static reactivity test data then predict an erosion rate of 0.040 to 0.100 inches per minute for the microcomposite (8Ta-55Hf-37C and 8Ta-54Hf-38C) as a result of the test firing. The erosion rate of the hypereutectic TaC-C composites is predicted to be in the range of 0.080 to 0.120 inches per minute. By such an analysis, a 60-second firing cycle would proceed for about 25 seconds with little or no erosion (perhaps about 0.1 mil per second). The remaining 35 seconds could yield an erosion rate as high as 2 mils per second in the case of the hypereutectic carbides. The average erosion rate observed for such a 60-second test would then be expected to be in the range of about 1.2 mils per second.

The prediction is based on the assumption that the limiting factor in erosion is the chemical reactions taking place between the nozzle insert material and the liquid alumina in the exhaust gases. The complexing effects of other species in the exhaust flame have not been assessed by the plasma jet or static reactivity laboratory tests, but they may well affect the overall erosion of the nozzle throat.

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13 ABSTRACT The magnitude of reactivity between $Al_2O_3$ and carbide composite rocket nozzle throat materials was assessed by means of laboratory tests. A secondary objective of the program was to establish the validity of the laboratory tests in evaluating material performance. The program included a plasma jet test that measured mechanical and chemical erosion and a static reactivity test to separate the purely chemical effects. A prediction of the performance of the carbide composites in a test firing was made, based on the laboratory tests.		

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